

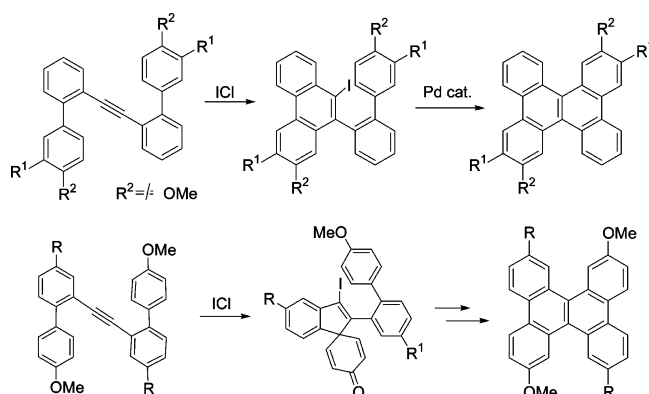
Synthesis of Dibenzo[*g,p*]chrysenes from Bis(biaryl)acetylenes via Sequential ICl-Induced Cyclization and Mizoroki–Heck Coupling

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We report a facile synthesis of functionalized dibenzo[*g,p*]chrysenes via initial ICl-promoted cyclization of bis(biaryl)acetylenes, followed by the Mizoroki–Heck coupling reaction. This new approach works well for various bis(biaryl)acetylenes to afford dibenzo[*g,p*]chrysenes bearing various functionalities. With substrates of one special type including 4'-methoxy-2-ethynylbiphenyls, we found that the ICl treatment led to *ipso* cyclization to give bicyclic spirocyclohexadienones. In the presence of MeOH/H₂SO₄, these spiroketone products undergo rearrangement to give 9-iodophenanthrenes through a selective 1,2-alkenyl migration. We prepared various 4'-methoxy-2-ethynylbiphenyl compounds to show the generalization of such an *ipso* cyclization and 1,2-alkenyl shift. This *ipso*-cyclization approach can be extended to the preparation of dibenzo[*g,p*]chrysenes.

Introduction

Polycyclic aromatic hydrocarbons have widespread applications as functional materials in various electronic devices such as nonlinear optical,¹ photo- and electroluminescent,² and molecule-based sensory devices.³ Dibenzo[*g,p*]chrysenes rep-

resent an interesting class of such compounds because of their attractive fluorescent properties such as high quantum yields, small Stoke shifts, and long-lived excited states.⁴ Synthesis of dibenzo[*g,p*]chrysenes is generally plagued by long procedures prior to the advent of SbCl₅/MeOH oxidation of bis(biaryl)-acetylenes by Swager.⁴ However, this method is limited to substrates of a special type (R¹ = OMe, eq 1) via the carbocation radical intermediate I. Iodine, ICl, and iodonium salt induced⁵ intramolecular acetylene cyclization is synthetically useful as it provides alkenyl iodides, which can be further functionalized

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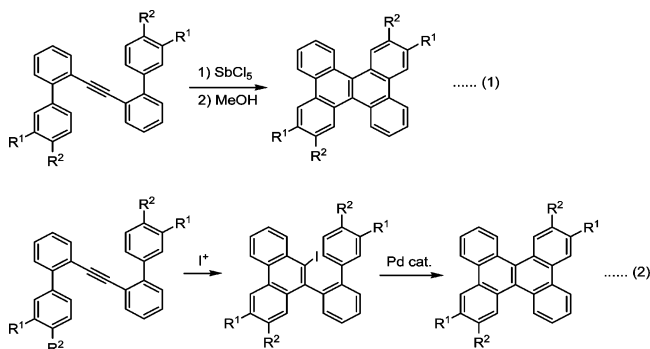
(1) (a) Munn, R. W.; Ironside, C. N. *Principles and Application of Nonlinear Optical Materials*; Chapman & Hall: London, UK, 1993. (b) Nie, W. *Adv. Mater.* **1993**, *5*, 520. (c) Prasad, P. N.; Williams, D. J. *Introduction to Nonlinear Optical Effects in Molecules and Polymers*; Wiley: New York, 1991.

(2) (a) Greenham, N. C.; Moratti, S. C.; Bradley, D. D. C.; Friend, R. H.; Holmes, A. B. *Nature* **1993**, *365*, 628. (b) Schwoerer, M. *Phys. Bull.* **1994**, *50*, 52. (c) Gruner, J.; Hamer, P. J.; Friend, R. H.; Huber, H. J.; Scherf, U.; Holmes, A. B. *Adv. Mater.* **1994**, *6*, 748. (d) Braun, D.; Heeger, A. J. *Appl. Phys. Lett.* **1991**, *58*, 1982.

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(4) Yamaguchi, S.; Swager, T. *J. Am. Chem. Soc.* **2001**, *123*, 12087.

by using palladium chemistry.⁶ To prepare dibenzo[*g,p*]chrysenes bearing diverse substituents, we have designed a two-step transformation involving an initial ICl-promoted cyclization to give iodophenanthrene **II**, followed by Pd-catalyzed cyclization, as depicted in eq 2. Here we report our findings on various dibenzo[*g,p*]chrysenes syntheses.



Results and Discussion

Bis(biaryl)acetylenes **1a–d** were prepared according to Swager's procedures.⁴ SbCl₅/MeOH oxidation was unsuccessful in preparing dibenzo[*g,p*]chrysenes from compounds **1a** and **1b** because of the lack of an activating group (R¹ = OMe). When species **1a** was subjected to ICl-induced^{7,8} cyclization, 9-iodophenanthrene **2a** was obtained in 90% yield via ortho cyclization. 3',5'-Difluoro-2-ethynylbiphenyl **1b** and 3',5'-dimethoxy-2-ethynylbiphenyl **1c** similarly afforded the corresponding iodophenanthrene derivatives **2b** and **2c** in 70% and 89% yields, respectively. This method was efficient for the cyclization of trimethoxy-substituted ethynyl biphenyl **1d**, as evident from the formation of 1,2,3-trimethoxy 9-iodophenanthrene **2d** in 95% yield. Although the R³ substituent of resulting product **2c** was also iodinated by ICl (R³ = I), this unexpected iodide was removable in the subsequent Pd-catalyzed cyclization (see Table 1).

When this cyclization was extended to 4'-methoxy-2-ethynylbiphenyls **3a,b**, as shown in Scheme 1, spiro[4.5]-cyclohexadienones **4a,b** were, however, obtained through *ipso*-cyclization in 87% and 88% yields, respectively; in these cases, we observed no corresponding phenanthrene derivatives. The distinct behavior of species **3a** and **3b** is attributed to the directing effect of the *p*-methoxy group to favor a 5-*endo-dig*

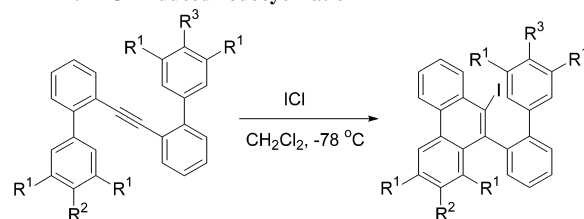
(5) For selected examples, see: (a) Barluenga, J.; Trincado, M.; Rubio, E.; Gonzalez, J. M. *Angew. Chem., Int. Ed.* **2003**, *42*, 2406. (b) Barluenga, J.; Gonzalez, J. M.; Campos, P. J.; Asensio, G. *Angew. Chem., Int. Ed.* **1988**, *27*, 1546. (c) Barluenga, J.; Llorente, I.; Alvarez-Garcia, L. J.; Gonzalez, J. M.; Compos, P. J.; Diaz, M. R.; Garcia-Granda, S. *J. Am. Chem. Soc.* **1997**, *119*, 6933. (d) Klein, T. R.; Bergemann, M.; Yehia, N. A. M.; Fanghanel, E. *J. Org. Chem.* **1998**, *63*, 4626. (e) Flynn, B. L.; Verdier-Pinard, P.; Hamel, E. *Org. Lett.* **2001**, *3*, 651. (f) Larock, R. C.; Yue, D. *Tetrahedron Lett.* **2001**, *42*, 6011. (g) Yue, D.; Larock, R. C. *J. Org. Chem.* **2002**, *67*, 1905. (h) Huang, Q.; Hunter, J. A.; Larock, R. C. *J. Org. Chem.* **2002**, *67*, 3437. (i) Arcadi, A.; Cacchi, S.; Di Giuseppe, S.; Fabrizi, G.; Marinelli, F. *Org. Lett.* **2002**, *4*, 2409. (j) Amjad, M.; Knight, D. W. *Tetrahedron Lett.* **2004**, *45*, 539.

(6) For selected reviews and examples, see: (a) Larock, R. C. *J. Organomet. Chem.* **1999**, *576*, 111. (b) Larock, R. C. *Palladium-Catalyzed Annulation in Perspectives in Organopalladium Chemistry for the XXI Century*; Tsuji, J., Ed.; Elsevier Press: Lausanne, Switzerland, 1999; pp 111–124. (c) Tsuji, J. *Palladium Reagents and Catalysts*; John Wiley & Sons: Chichester, UK, 1999. (d) Arcadi, A.; Cacchi, S.; Fabrizi, G.; Moro, L. *Eur. J. Org. Chem.* **1999**, 1137. (e) Campo, M. A.; Larock, R. C. *J. Org. Chem.* **2002**, *67*, 5616.

(7) Yao, T.; Campo, M. A.; Larock, R. C. *Org. Lett.* **2004**, *6*, 2677.

(8) Yao, T.; Campo, M. A.; Larock, R. C. *J. Org. Chem.* **2005**, *70*, 3511.

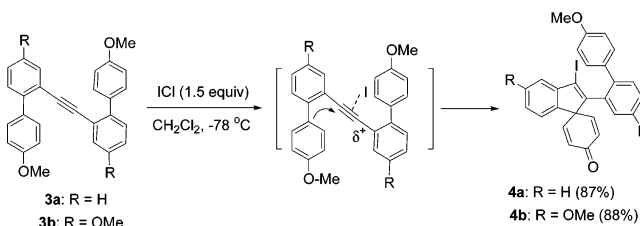
TABLE 1. ICl-Induced Iodocyclization



entry	ethynylarenes ^a	iodoarenes (% yield ^b)
1	1a (R ¹ = R ² = R ³ = H)	2a (R ¹ = R ² = R ³ = H, 90%)
2	1b (R ¹ = H, R ² = R ³ = F)	2b (R ¹ = H, R ² = R ³ = F, 70%)
3	1c (R ¹ = OMe, R ² = R ³ = H)	2c (R ¹ = OMe, R ² = H, R ³ = I, 89%)
4	1d (R ¹ = R ² = R ³ = OMe)	2d (R ¹ = R ² = R ³ = OMe, 95%)

^a [Substrate] = 0.1 M, ICl (1.5 equiv for entries 1, 2, and 4 and 2.5 equiv for entry 3). ^b Yields are given after column chromatography on a silica column.

SCHEME 1



cyclization. ICl-induced *ipso*-cyclization was reported for 4-(*p*-methoxyaryl)alk-1-yne by Larock.^{8,9} As spiro structural motifs are present frameworks in some natural products,¹⁰ as well as some organic optoelectronic materials,¹¹ we proceeded to investigate this ICl-induced spiro cyclization in detail.

We prepared various 4'-methoxy-2-alkynylbiphenyl compounds **5a–n** to examine the generality of this ICl-induced cyclization; our results are depicted in Table 2. Entries 1–8 show the variation of the alkynyl R substituent of substrates with phenyl (R = C₆H₅, 4-MeC₆H₄, and 4-MeOC₆H₄), heteroaryl (R = 2-furyl, 2-thienyl, 2-benzofuryl, and 2-benzo[*b*]-thienyl), and alkyl (R = *n*-C₆H₁₃) groups; the corresponding spirocyclohexadienones were obtained in yields exceeding 88%. Here we obtained no byproduct stemmed from iodination of the reactive furan or thiophene moieties. This ICl-induced *ipso*-cyclization also worked efficiently for substrates **5i–n** bearing methoxy at the bridging phenyl C(4) and C(5) positions; the desired spirocyclohexadienones **6i–n** were obtained in 88–94% yields. The crystal structure of compound **6j** was confirmed by an X-ray diffraction study.¹²

We also prepared 2',4'-dimethoxy-2-alkynylbiphenyl species **7**, which upon treatment with ICl produced 9-iodophenanthrene

(9) For ICl-induced *ipso*-cyclization, see: (a) Appel, T. R.; Yehia, N. A. M.; Baumeister, U.; Hartung, H.; Kluge, R.; Strohl, D.; Fanghanel, E. *Eur. J. Org. Chem.* **2003**, *47*. (b) Zhang, X.; Larock, R. C. *J. Am. Chem. Soc.* **2005**, *127*, 12230.

(10) (a) For a review, see: Heathcock, C. H.; Graham, S. L.; Pirrung, M. C.; Plavac, F.; White, C. T. In *The Total Synthesis of Natural Products*; Apsimon, J., Ed.; Wiley-Interscience: New York, 1983; Vol. 5, pp 264–313. (b) Sakamoto, K.; Tsuji, E.; Abe, F.; Nakanishi, T.; Yamashita, M.; Shigematsu, N.; Izumi, S.; Okuhara, M. *J. Antibiot.* **1996**, *49*, 37. (c) Gonzalez-Lopez de Turiso, F.; Curran, D. P. *Org. Lett.* **2005**, *7*, 151 and references cited therein.

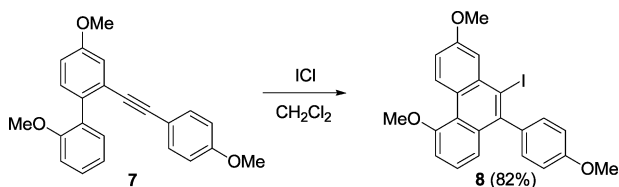
(11) Saragi, T. P. I.; Spehr, T.; Siebert, A.; Fuhrmann-Lieker, T.; Salbeck, J. *Chem. Rev.* **2007**, *107*, 1011.

(12) X-ray data for compound **6j** are provided in the Supporting Information.

TABLE 2. ICI-Induced Synthesis of Spirocyclohexadienone Derivatives and Their Acid-Promoted Rearrangements

alkynylarenes ^a	spiro compounds ^b (% yield) ^c	arenes (Yield%) ^c
(1) 5a : R = Ph	6a (92%)	9a (92%)
(2) 5b : R = 4-MeC ₆ H ₄	6b (94%)	9b (94%)
(3) 5c : R = 4-MeOC ₆ H ₄	6c (95%)	9c (93%)
(4) 5d : R = 2-furan	6d (91%)	9d (92%)
(5) 5e : R = 2-thiophene	6e (93%)	9e (93%)
(6) 5f : R = 2-benzofuran	6f (95%)	9f (95%)
(7) 5g : R = 2-benzo[<i>b</i>]thiophene	6g (94%)	9g (94%)
(8) 5h : R = <i>n</i> -C ₆ H ₁₃	6h (88%)	9h (89%)
(9) 5i : R = Ph	6i (91%)	9i (94%)
(10) 5j : R = 4-MeOC ₆ H ₄	6j (90%)	9j (95%)
(11) 5k : R = 2-furan	6k (89%)	9k (94%)
(12) 5l : R = 2-thiophene	6l (92%)	9l (94%)
(13) 5m : R = <i>n</i> -C ₆ H ₁₃	6m (88%)	9m (90%)
(14) 5n : R = 4-MeOC ₆ H ₄	6n (94%)	9n (94%)

^a 0.1 M [substrate] in CH₂Cl₂, -78 °C for 2 h. ^b 0.1 M [substrate] in CH₂Cl₂/MeOH (1:1, v/v), H₂SO₄ (2.0 equiv), 0 to 23 °C, 6 h. ^c Yields are given after purification by column chromatography on silica gel.

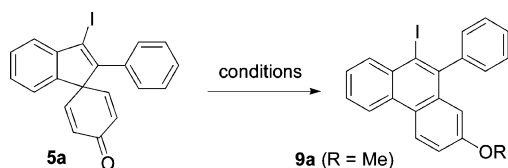
SCHEME 2

8 in an 82% yield (Scheme 2). The *ipso*-chemoselectivity is apparently limited to substrates bearing a *p*-methoxy-substituted phenyl ring.¹³

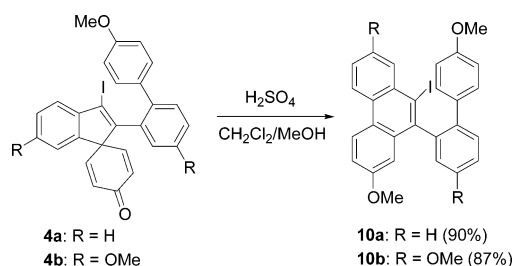
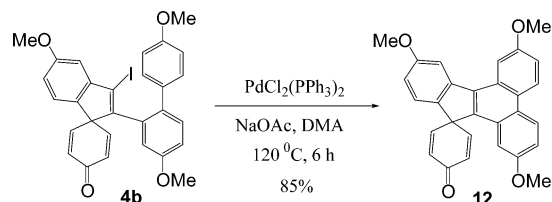
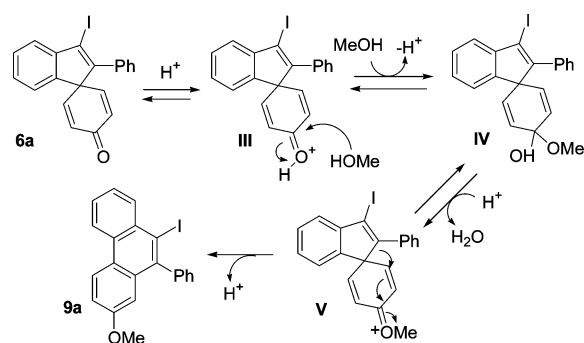
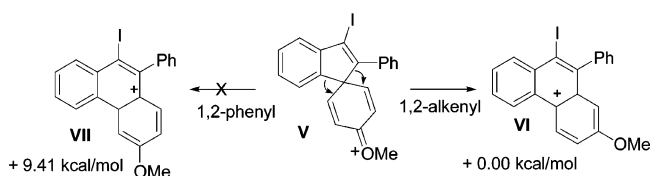
The preceding spirocyclohexadienones **6a–n** are expected to undergo isomerization to give iodophenanthrene products through the dienone–phenol rearrangement, generally catalyzed by Brønsted acids.¹⁴ As shown in Scheme 3, the organic acid *p*-TSA (10 mol %) or H₂SO₄ (2.0 equiv) gave no rearranged products even under refluxing in CH₂Cl₂ for 12 h. To our delight, the use of H₂SO₄ (2.0 equiv) in mixed CH₂Cl₂/MeOH (1:1, v/v) at 23 °C for 2 h afforded rearranged product **9a** in a 93% yield. The structure of compound **9a** was carefully elucidated from its ¹H NOE spectrum, which reveals that the acid-catalyzed rearrangement proceeds via a 1,2-migration of the alkenyl rather than the phenyl group. Scheme 4 shows an

(13) We do not exclude the possibility that *ipso*-cyclization actually occurred for species **7** bearing an *o*-methoxy group, which underwent a rapid rearrangement under the reaction conditions.

(14) (a) Shine, H. J. In *Aromatic Rearrangements*; Elsevier: New York, 1967; pp 55–68. (b) Schultz, A. G.; Hardinger, S. A. *J. Org. Chem.* **1991**, *56*, 1105. (c) Zimmerman, H. E.; Cirkva, V. *J. Org. Chem.* **2001**, *66*, 1839.

SCHEME 3

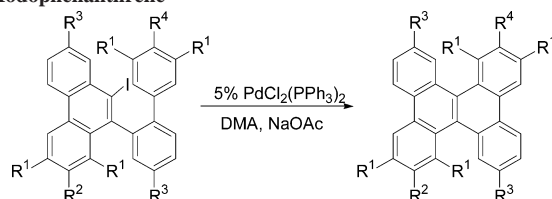
solvent	condition	products
(1) CH ₂ Cl ₂	<i>p</i> -TSA (10%) reflux, 12 h	NR
(2) CH ₂ Cl ₂	H ₂ SO ₄ (2.0 equiv.), reflux, 12 h	NR
(3) CH ₂ Cl ₂ /MeOH (1:1, v/v)	H ₂ SO ₄ (2.0 equiv.), rt, 6 h	9a : R = Me (93%)

SCHEME 4**SCHEME 5****SCHEME 6****SCHEME 7**

extension of this rearrangement to species **4a** and **4b**, which gave products **10a** and **10b** in 90% and 87% yields, respectively. Notably, the original protocol in eq 2 would have provided the same products **10a** and **10b** if the ICI-induced 6-*endo-dig* cyclization were applicable to initial substrates **3a** and **3b**.

Using these optimum conditions, we further examined the generality of this acid-catalyzed 1,2-alkenyl migration by treating the remaining spirocyclohexadienones **6b–n** with H₂

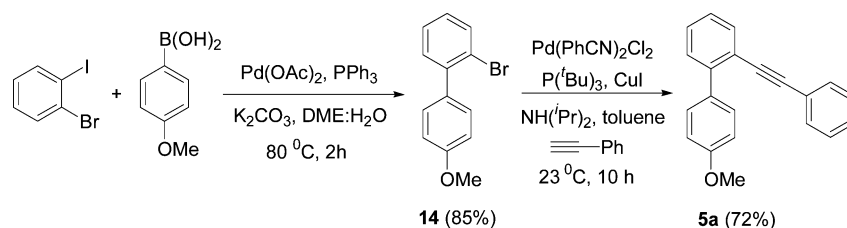
TABLE 3. Intramolecular Arylation of Iodophenanthrene



entry	ethynylarenes ^a	iodoarenes (% yield ^b)
1	2a (R ¹ = R ² = R ³ = R ⁴ = H)	11a (R ¹ = R ² = R ³ = R ⁴ = H, 71%)
2	2b (R ¹ = R ³ , R ² = R ⁴ = F)	11b (R ¹ = R ³ = H, R ² = R ⁴ = F, 72%)
3	2c (R ¹ = OMe, R ² = R ³ = H, R ⁴ = I)	11c (R ¹ = OMe, R ² = R ³ = R ⁴ = H, 69%)
4	2d (R ¹ = R ² = R ⁴ = OMe, R ³ = H)	11d (R ¹ = R ² = R ⁴ = OMe, R ³ = H, 68%)
5	10b (R ¹ = H, R ² = R ³ = R ⁴ = OMe)	11e (R ¹ = H, R ² = R ³ = R ⁴ = OMe, 85%)

^a 0.05 M [substrate] in DMA, at 120 °C for 4 h with 5% PdCl₂(PPh₃)₂ and NaOAc (2.0 equiv). ^b Yields are given after purification by column chromatography on silica gel.

SCHEME 8



SO₄ in MeOH/CH₂Cl₂; the results are summarized in Table 2. The rearrangement occurred efficiently to generate the desired 9-iodophenanthrene products **9b–n**, with yields exceeding 89%. In these cases, we obtained no side product corresponding to a 1,2-phenyl migration. The diversity of substrates used demonstrates the scope and reliability of this rearrangement. The ¹H NMR spectral signals of all these products have a common singlet signal in the upfield region (δ 6.75–7.05 ppm), characteristic of the phenanthrene C-(1) proton resonance.

As shown in Table 3, we sought to accomplish the ultimate synthesis of dibenzo[*g,p*]chrysenes using 9-iodophenanthrenes **2a–d** and **10a,b**, via two distinctly different pathways, with the second cyclization being based on an intramolecular Mizoroki–Heck coupling.¹⁵ Treatment of iodoarene **2a** with PdCl₂(PPh₃)₂ (5 mol %) and sodium acetate (2.0 equiv) in DMA (*N,N*-dimethylacetamide) at 120 °C for 4 h afforded the desired derivative **11a** in a 71% yield. This set of conditions was extendible to other 9-iodophenanthrenes **2b**, **2d**, and **10b**, giving the desired products **11b** and **11d,e** in 68–85% yields. Notably, the extra iodide (R⁴ = I) of species **2c** was reduced in this cyclization to afford product **11c** (69%). Scheme 5 shows an alternative application of these spiroketone products; treatment of alkenyl species **4b** with catalyst PdCl₂(PPh₃)₂ and NaOAc in DMA afforded the cyclized product **12** in an 85% yield.

Scheme 6 depicts a plausible mechanism for the rearrangement of spirodienone **6a** to 9-iodophenanthrene **9a** promoted by MeOH/H₂SO₄. In this transformation, a proton initially enhances the addition of methanol at the ketone group of species **6a** to form species **IV**, which ultimately forms species **V** in the presence of excess MeOH. In contrast with species **III**, the methoxy group of species **V** is kinetically stable and stabilizes the oxonium center to induce a 1,2-alkenyl migration. We cannot

exclude the possibility that because MeOH is more polar than CH₂Cl₂ it facilitates this migration.

As depicted in Scheme 7, we performed a calculation to estimate the relative stability of two carbocations **VI** and **VII**, corresponding to the 1,2-migration of an alkenyl and a phenyl group, respectively. According to the B3LYP/6-31+G* calculation (3-21G* for iodine atom since 6-31+G* is not available), carbocation **VI** has +9.41 kcal/mol less in energy than species **VII**; this result supports a 1,2-alkenyl migration.

Conclusion

We have achieved a facile synthesis of functionalized dibenzo[*g,p*]chrysenes via initial ICl-promoted cyclization of bis(biaryl)acetylenes, followed by Pd-catalyzed annulation of the resulting 9-iodophenanthrene intermediates. For substrates of one particular type, such as 4'-methoxy-2-ethynylbiphenyls, we found that this ICl treatment leads to *ipso* cyclization to give bicyclic spirocyclohexadienones, which are potentially useful intermediates in organic synthesis. In the presence of MeOH/H₂SO₄, these spirocyclohexadienones undergo rearrangement to give 9-iodophenanthrenes through a selective 1,2-migration of the alkenyl group. The feasibility of this migration is supported by calculation.

Experimental Sections

(1) Representative Synthesis of 2-Ethynyl Biphenyls: Synthesis of 4'-Methoxy-2-(phenylethynyl)biphenyl (**5a**). (a) Synthesis of 2-Bromo-4'-methoxybiphenyl (**14**). To a solution of Pd(OAc)₂ (101 mg, 0.45 mmol), PPh₃ (354 mg, 1.35 mmol), and K₂CO₃ (2.76 g, 20.0 mmol) in DME (30 mL) and H₂O (10 mL) near 23 °C was added 4-methoxyphenylboronic acid (1.67 g, 11.0 mmol);^{16,17} the resulting mixture was stirred for 10 min. To this

(15) (a) Ames, D. E.; Opalko, A. *Tetrahedron* **1984**, *40*, 1919. (b) For a recent review on aryl–aryl bond formation, see: Alberico, D.; Scott, M. E.; Lautens, M. *Chem. Rev.* **2007**, *107*, 174.

(16) Karig, G.; Moon, M.-T.; Thasana, N.; Gallagher, T. *Org. Lett.* **2002**, *4*, 3115.

solution was added 2-iodo-1-bromobenzene (2.83 g, 10.0 mmol), and the mixture was heated to 80 °C for 2 h. The resulting solution was cooled to 23 °C before addition of saturated ammonium chloride solution and extraction with ethyl acetate. The extracts were washed with brine solution, dried over MgSO₄, filtered, and concentrated. The mixtures were eluted through a silica column to give bromobiaryl species **14** (2.24 g, 8.5 mmol, 85%) as a yellow oil.

(b) Synthesis of 4'-Methoxy-2-(phenylethynyl)biphenyl (5a). To Pd(PPhCN)₂Cl₂ (195 mg, 0.509 mmol) and CuI (99 mg, 0.509 mmol) was added dry toluene (20 mL), P(*t*-Bu)₃ (206 mg 1.02 mmol), HN(*i*-Pr)₂ under N₂, and bromo derivative **14** (2.24 g, 8.5 mmol), and the mixture was stirred at 23 °C before addition of phenyl acetylene (1.23 g, 12.21 mmol). After 10 h, the reaction mixture was treated with hexane (40 mL), filtered through a small silica pad, concentrated, and purified by flash chromatography, which yielded the desired product **5a** (1.74 g, 72%) as a viscous oil. Spectral data for compound **5a**: ¹H NMR (CDCl₃, 400 MHz) δ 7.60–7.62 (m, 3H), 7.34–7.40 (m, 4H), 7.26–7.28 (m, 4H), 6.98 (dd, 2H, *J* = 8.8, 0.8 Hz), 3.86 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 159.1, 143.4, 133.0, 132.9, 131.3, 130.5, 129.3, 128.5, 128.2, 128.0, 126.6, 123.5, 121.4, 113.3, 92.1, 89.6, 55.3; HRMS (EI) calcd for C₂₁H₁₆O (M⁺) 284.1201, found 284.1209.

(2) Representative Procedure for ICl-Induced *Ips*o Cyclization: Synthesis of Spirocyclohexadienone 6a. A solution of compound **5a** (500 mg, 1.76 mmol) in dry CH₂Cl₂ (18 mL) was maintained at –78 °C with an acetone–liquid N₂ bath. To this solution was added ICl (3.51 mL, 1 M solution in CH₂Cl₂), using a standard syringe. The reaction was maintained at the same temperature, and monitored by TLC. After the completion of the reaction, it was quenched with a saturated sodium thiosulfate solution and warmed to 22 °C. The reaction mixture was extracted with CH₂Cl₂ (2 × 20 mL); the combined extracts were washed with brine solution, dried over MgSO₄, concentrated, and chromatographed through a silica column to yield product **6a** (642 mg, 1.62 mmol, 92%) as a yellow solid (mp 206.3–208.5 °C). Spectral data for compound **6a**: ¹H NMR (CDCl₃, 400 MHz) δ 7.44 (d, 2H, *J* = 2.4 Hz), 7.29–7.33 (m, 6H), 7.08 (d, 1H, *J* = 7.6 Hz), 6.41–6.50 (m, 4H); ¹³C NMR (CDCl₃, 100 MHz) δ 185.7, 150.0, 147.0, 145.8, 140.0, 134.7, 131.0, 129.1, 128.7, 128.3, 128.2, 127.8, 123.8, 123.2, 99.8, 62.3; HRMS (EI) calcd for C₂₀H₁₃IO (M⁺) 396.0011, found 396.0007.

(3) Representative Procedure for Acid-Mediated Rearrangements: Synthesis of Iodophenanthrene 9a. Compound **6a** (500 gm, 1.26 mmol) was dissolved in a mixture of CH₂Cl₂ and MeOH (12.5 mL, v/v, 1:1) and cooled to 0 °C with an ice water bath. At this temperature, H₂SO₄ (247 mg, 2.52 mmol) was added dropwise with use of a glass dropper for 10 min; the reaction was warmed

to 22 °C, stirred for 6 h, and monitored with thin-layer chromatography. After completion, the reaction mixture was treated with water and extracted with CH₂Cl₂ (3 × 20 mL). The organic layers were washed with brine solution and dried over MgSO₄. The organic layer was filtered, concentrated, and purified by using column chromatography, which delivered the product **9a** (481 mg, 1.17 mmol, 93%) as a yellow solid (mp 137.2–138.5 °C). Spectral data for compound **9a**: ¹H NMR (CDCl₃, 400 MHz) δ 8.58 (d, 1H, *J* = 8.8 Hz), 8.52 (d, 1H, *J* = 8.0 Hz), 8.42 (d, 1H, *J* = 8.0 Hz), 7.49–7.64 (m, 5H), 7.24–7.28 (m, 3H), 6.75 (d, 1H, *J* = 2.4 Hz), 3.66 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 158.2, 145.4, 144.7, 134.6, 133.7, 131.4, 130.5, 129.8, 128.5, 127.8, 127.5, 127.0, 124.6, 124.1, 122.1, 116.8, 109.7, 109.1, 55.1; HRMS (EI) calcd for C₂₁H₁₅IO (M⁺) 410.0168, found 410.0166.

(4) Representative Procedure for ICl-Induced Cyclization: Synthesis of Iodophenanthrene 8. Compound **8** was obtained in a 82% yield by treatment of biphenyl species **7** with ICl in cold CH₂Cl₂ (–78 °C); the procedure was similar to that of species **6a**. Yellow solid (mp 112.3–113.5 °C); IR (neat, cm^{–1}) 3075, 3025, 2832, 1607, 1243; ¹H NMR (CDCl₃, 400 MHz) δ 9.64 (d, 1H, *J* = 9.6 Hz), 7.96 (d, 1H, *J* = 2.8 Hz), 7.24–7.29 (m, 2H), 7.12–7.17 (m, 3H), 7.04–7.05 (m, 3H), 4.11 (s, 3H), 4.00 (s, 3H), 3.91 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 159.1, 158.6, 157.7, 145.5, 139.2, 134.5, 133.9, 131.0, 130.5, 125.6, 124.8, 121.9, 121.2, 116.8, 115.8, 113.8, 109.1, 108.8, 56.0, 55.4, 55.3; HRMS calcd for C₂₃H₁₉IO₃ (M⁺) 470.0379, found 470.0375.

(5) Representative Procedure for Intramolecular Mizoroki–Heck Coupling: Synthesis of Dibenzo[*g,p*]chrysenes 11a. A two-necked flask was charged with iodophenanthrene **2a** (200 mg, 0.44 mmol), PdCl₂(PPh₃)₂ (15 mg, 0.022 mmol), and CH₃COONa (72 mg, 0.88 mmol) under a nitrogen atmosphere, followed by addition of DMA (9 mL). The resulting solution was heated at 120 °C for 4 h, then cooled to 22 °C and HCl solution (100 mL, 1 M) was added. The resulting mixture was extracted with dichloromethane, and the organic layers were combined, dried over MgSO₄, and concentrated under reduced pressure. The residue was chromatographed on a silica column to afford **11a** (102 mg, 0.31 mmol, 71%) as a yellow solid (mp 210.5–213.4 °C). Spectral data for compound **11a**: ¹H NMR (CDCl₃, 400 MHz) δ 8.68–8.70 (m, 8H), 7.60–7.69 (m, 8H); ¹³C NMR (CDCl₃, 100 MHz) δ 130.7, 129.1, 128.8, 127.4, 126.4, 123.5 (one peak merged); HRMS (EI) calcd for C₂₆H₁₆ (M⁺) 328.1252, found 328.1256.

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Supporting Information Available: Experimental procedures for the preparation of **1a**, X-ray data of compound **6j**, spectral data, and copies of NMR spectra of key compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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